

# N<sub>2</sub> influence on the vibrational distribution of the asymmetric level of CO<sub>2</sub>

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**Abstract:** This work contributes towards a detailed kinetic model to study the plasma chemistry in CO<sub>2</sub>-N<sub>2</sub> plasmas, in order to explore the possibility of admixing nitrogen to enhance CO<sub>2</sub> dissociation. A particular interest is dedicated to the pumping of the asymmetric vibrational mode of CO<sub>2</sub>, considered as a promising way for the CO<sub>2</sub> dissociation at lower energy costs than by direct electronic impact. For this purpose, 0-D simulations are performed reproducing the conditions of the experiments lead in parallel at LPP. The input data of the simulations must comprise sets of reaction rates for the vibrational-translational exchanges (VT) and the vibrational-vibrational exchanges (VV) between N<sub>2</sub>, CO<sub>2</sub> and the derived products. This presentation focus on the computation and the validation of these data sets and the first results obtained from a simplified kinetic model.

## 1. Introduction

Carbon dioxide dissociation is a way of both reducing one of the main greenhouse gases and providing a source of synthetic fuels, avoiding the carbon footprint and the restructuring problems linked to energy distribution. Non-equilibrium plasmas seem to be a very interesting medium to efficiently dissociate the CO<sub>2</sub> molecule, with reported energy efficiencies of about 45% under industrial conditions [1, 2]. This work explores the possibilities of admixing nitrogen to the CO<sub>2</sub> plasma to enhance the dissociation efficiency. The aim is to develop a kinetic model to study the plasma chemistry in CO<sub>2</sub>-N<sub>2</sub> plasmas.

## 2. Discussion

The VT and VV reaction rates between CO<sub>2</sub> and N<sub>2</sub> molecules have been computed for the first vibrational levels, using a temperature power law which fits the results from previous experiments [3]. For the upper levels these rates have been scaled using the SSH theory [4]. In total, vibrational levels up to  $v \leq 5$  for N<sub>2</sub>, and  $v_1 \leq 3$ ,  $v_2 \leq 6$ ,  $v_3 \leq 5$  for the symmetric stretching, bending and asymmetric stretching modes of CO<sub>2</sub>, respectively, are taken into account in the model.

Figure 1 shows the time evolution of the population of the CO<sub>2</sub>(00<sup>0</sup>1) level in the afterglow of a dc discharge for  $p=5$  Torr. The gas temperature profile in the afterglow and the initial densities of CO<sub>2</sub> molecules in the different vibrational modes are taken from experiments in a pure CO<sub>2</sub> dc discharge at a current  $I=50$  mA. The results confirm the potential of N<sub>2</sub> addition to enhance the vibrational pumping of the CO<sub>2</sub> asymmetric mode, whenever the characteristic vibrational temperature of N<sub>2</sub> is larger than that of the CO<sub>2</sub> asymmetric mode.

## 3. References

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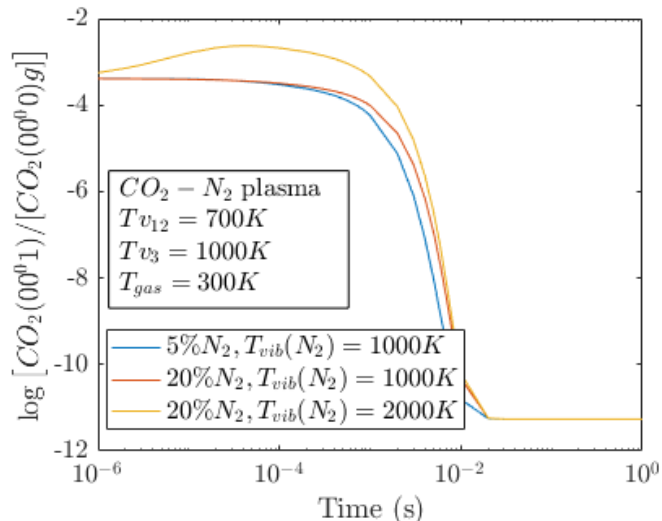


Fig.1: Time evolution of the population of CO<sub>2</sub>(00<sup>0</sup>1) level in a CO<sub>2</sub>-N<sub>2</sub> afterglow, at 5 Torr, for different N<sub>2</sub> concentrations and vibrational temperatures.

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