

Modelling of N₂-H₂ capacitively coupled radio-frequency discharges

M. Jiménez-Redondo¹, L. Marques^{1,2}, N. Carrasco³, G. Cernogora³, L. L. Alves²

¹*Centro de Física das Universidades do Minho e do Porto, Universidade do Minho, 4710-057, Braga, Portugal*

²*Instituto de Plasmas e Fusão Nuclear, Instituto Superior Técnico, Univ. Técnica de Lisboa, Lisboa, Portugal*

³*Université Versailles St-Quentin, CNRS, LATMOS, 11 blvd d'Alembert, 78280 Guyancourt, France*

In this work, we present the results of simulations carried out for N₂-H₂ low pressure, low power capacitively coupled radio-frequency discharges, for amounts of H₂ up to 5%. Simulations are performed using a hybrid code that couples a two-dimensional time-dependent fluid module, describing the dynamics of the charged particles, to a zero-dimensional kinetic module, that solves the Boltzmann equation and describes the production and destruction of neutral species. The model accounts for the production of several excited states, and contains a detailed surface chemistry that includes recombination processes and the production of NH_x molecules. Simulations show that surface production of NH₃ plays a key role in the neutral and ion kinetics of the discharge.

1. Introduction

Capacitively coupled radio-frequency (ccrf) discharges in nitrogen-containing mixtures have been used in planetary studies to simulate, in laboratory environment, the reactivity of ionospheres. The present work is part of a research strategy, involving both simulations and experiment, to analyse the N₂-CH₄ ionospheric chemistry of Titan, the biggest satellite of Saturn. The first step was the study of ccrf discharges in pure N₂ [1], and now continues with the analysis of N₂-H₂ discharges.

2. Modelling

The simulations run at low pressures (0.6–1.2 mbar), for 30–100 sccm gas flows and 5–20 W coupled powers, in N₂-H₂ mixtures with hydrogen concentrations up to 5%.

The model consists of a hybrid code that couples a two-dimensional (r, z) time-dependent fluid-type module, which describes the transport of the charged particles, to a very complete zero-dimensional kinetic module for the nitrogen-hydrogen mixture. The fluid module solves the continuity and the momentum transfer equations for electrons, positive ions N₂⁺, N₄⁺, H⁺, H₂⁺, H₃⁺, HN₂⁺, NH⁺, NH₂⁺, NH₃⁺ and NH₄⁺, and negative ions H⁻ and NH₂⁻, the electron mean energy transport equations, and Poisson's equation for the rf electric potential. The space-time map of the electron transport and rate coefficients are obtained from the electron mean energy profile, using the local mean energy approximation [1,2]. The kinetic module solves the two-term homogeneous and stationary electron Boltzmann equation (accounting for inelastic collisions from ground-state molecules and atoms, and inelastic and superelastic collisions

involving vibrationally excited states) and the rate balance equations of the ground-state vibrational excited states of N₂ and H₂, the most relevant electronic excited states for N₂ and N, H, and the most important crossed-species NH_y (y=1-3) and N₂H_y (y=2-4) resulting from interactions within the N₂/H₂ systems [2,3]. An extended surface chemistry is considered, taking in account adsorption, surface association and heterogeneous reactions, which are key to the formation of NH₃. The electron impact chemistry of hydrogen has been updated using the latest set of cross sections available from the IST-Lisbon database of LXCat [4,5].

3. Results

Simulations show that significant amounts of NH₃ are produced in the discharge, with a high dependence on the parameters for the surface kinetics. The abundances of positive ions are greatly affected by the neutral composition, with NH₄⁺ quickly becoming the major ion when sufficient NH₃ is present.

4. Acknowledgments

This work was partially supported by the Portuguese FCT under project UID/FIS/50010/2013.

5. References

- [1] L.L. Alves et al, Plasma Sources Sci. Technol. **21** (2012) 045008.
- [2] L. Marques, J. Jolly, L.L. Alves, J. Appl. Phys. **102** (2007) 063305.
- [3] E. Tatarova et al, Plasma Sources Sci. Technol. **14** (2005) 19.
- [4] L.L. Alves, J. Phys. Conf. Ser. **565** (2014) 012007.
- [5] www.lxcat.net